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Dynamics of Ion Beam Stimulated Surface Mass Transport to Nanopores

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ABSTRACT

We explore the ion beam-induced dynamics of the formation of large features at the edges of nanopores in freestanding silicon nitride membranes. The shape and size of these “nanovolcanoes”, together with the rate at which the nanopores open or close, are shown to be strongly influenced by sample temperature. Volcano formation and pore closing slow and stop at low temperatures and saturate at high temperatures. Nanopore volcano size and closing rates are dependent on initial pore size. We discuss both surface diffusion and viscous flow models in the context of these observed phenomena.

INTRODUCTION

The advent of solid state nanopores as single-molecule detectors [1] has highlighted the importance of characterizing novel methods for fabricating nanostructures. Ion beam sculpting [2] is a robust method for making solid state nanopores, but the processes involved in the fabrication process remain poorly understood. While many potentially relevant processes, such as surface diffusion [2], sputtering [3–4], anisotropic deformation via thermal spikes [5–6], ion-enhanced viscous flow [6–8], radiation-induced stresses [9], and combinations of the above [10] have been studied with MeV beams or with crystalline targets, there is little data available regarding interactions of lower-energy keV beams with amorphous materials such as those found widely in silicon-based technology. In particular, it is unclear how matter transport occurs over large length scales when irradiated with low-energy beams that deposit their energy within only a few nanometers of the surface.

In this paper, we discuss several previously unreported features of nanopore fabrication using keV ion beams and discuss their ramifications with respect to various extant models for the observed matter transport.

EXPERIMENT

Samples were fabricated from low-stress amorphous silicon nitride grown by low-pressure chemical vapor deposition (LPCVD) on a silicon substrate. The LPCVD was performed at Cornell’s Nanofabrication Facility and yields silicon nitride thin films with a nominal tensile stress of 180 MPa [11]. The stoichiometry is $\text{Si}_{3.5}\text{N}_4$, as determined by Rutherford backscattering. Freestanding square membranes 90 μm on a side and 500 nm thick were produced by lithographic patterning and a subsequent anisotropic etch of the underlying silicon substrate in KOH.

Starting holes of the desired diameter were drilled at the center of the freestanding membrane, from the substrate side, by rastering a 10-nm diameter, 50 keV gallium ion beam produced by a FEI/Micrion 9500 focused ion beam (FIB) instrument. The resulting holes were conical in shape, opening towards the substrate side of the membrane with an apex angle of about 10 degrees, as determined by atomic force microscopy (AFM).

After fabrication, samples were stored in a nitrogen-purged dry box until used. To ensure consistency of results, samples not in the dry box for more than a day were stored in methanol or ethanol and blown dry with dry nitrogen immediately before use. Samples were sometimes transferred directly from the FIB to the sculpting chamber, in which case no cleaning procedures were employed.

Ion beam sculpting experiments that form “nanovolcanoes” and modify the size of the nanopore were run in a home-built ion sculpting apparatus described elsewhere [12]. The base pressure of the sculpting chamber was about 7×10^{-10} torr with a cold can surrounding the sample in operation. The sculpting beam was a 3 keV ion beam of Ar^+ with a spot size of ~ 0.2 mm, which is significantly larger than the freestanding silicon nitride membrane upon which it impinges. The changing nanopore hole size was monitored instantaneously during the sculpting process with a channeltron single-ion detector located behind the sample (see Figure 1a for a highly schematic view).

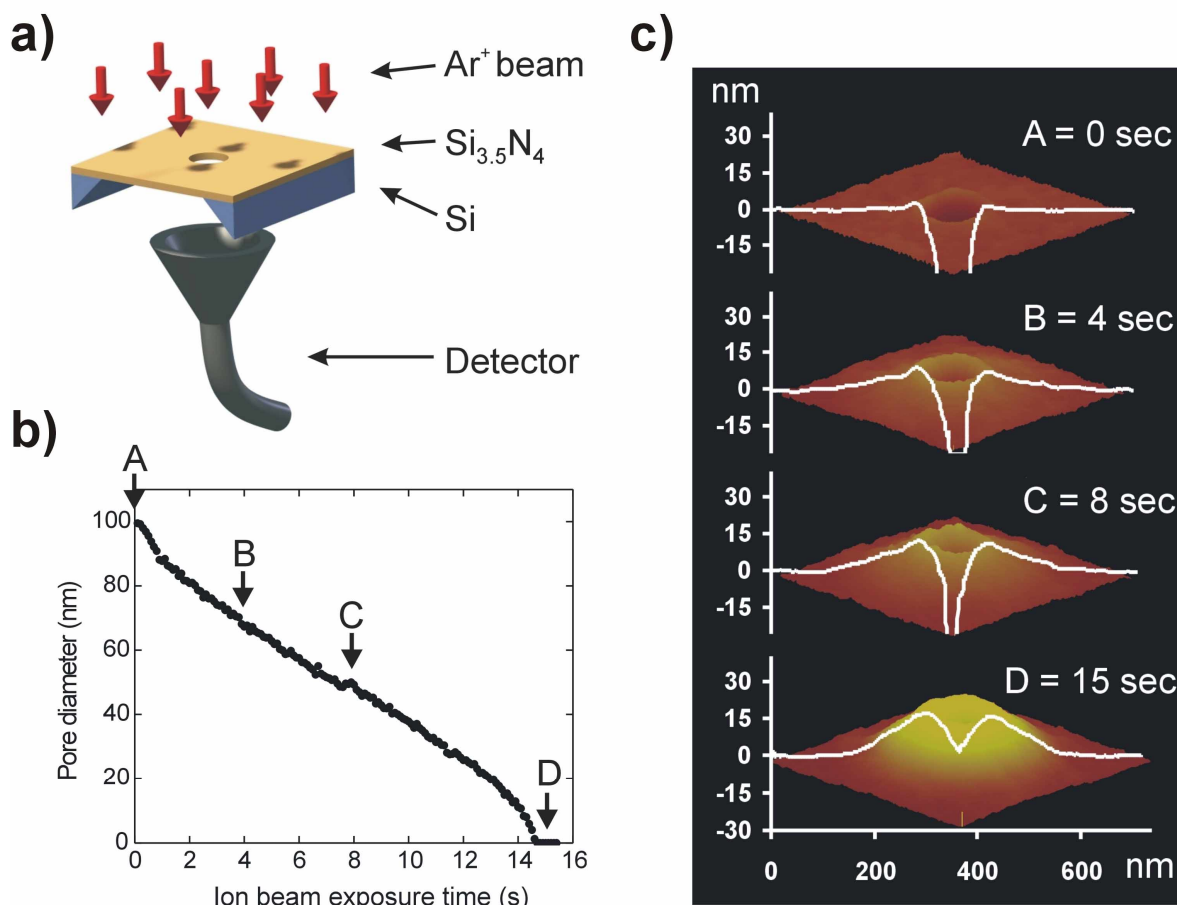


Figure 1. (a) Representation of ion sculpting experimental setup. (b) Pore shrinkage with time as deduced from the count rate. (c) Nanovolcano evolution.

To ensure consistency of results, each sample was baked at 80 °C for 5 minutes in the sculpting chamber prior to ion beam alignment. The beam was then aligned with the pore at a temperature of -30 °C, which is below the temperature at which most silicon nitride nanopores begin to close [2].

After ion beam sculpting the nanopores under the desired experimental conditions and recording the instantaneously changing rates at which ions were transmitted through the pore during the process, we studied the samples with an atomic force microscope. All topography analysis was performed on a Digital Instruments 3100 Nanoscope IIIa atomic force microscope (AFM) operating in tapping mode. Tips were Mikromasch NSC35A or NSC35B, with nominal tip radii of less than 10 nm and spring constants 7.5 N/m and 14 N/m, respectively. Images were processed with WSxM software [13].

EXPERIMENTAL RESULTS

A typical nanopore closing curve and volcano formation observed at different stages of the pore closing process are shown in Figure 1b-c. This pore was sculpted at 28 °C and at a flux of 4.4 Ar⁺/nm²/sec. The transmitted flux through the pore is monitored by the channeltron detector (Figure 1a), producing the diameter vs. time closing curve of Figure 1b. At various stages of closing, the sample was removed from the sputtering chamber and probed with an AFM, yielding the topographs in Figure 1c [14].

Temperature dependence

We have observed that both the instantaneous closing rate (*i.e.* the change in nanopore area with fluence or time, or the slope of an area vs. fluence curve) and volcano size vary significantly with sample temperature. Figure 2 depicts volcano shapes at varying temperature. All pores were closed from the same initial diameter (~ 100 nm) to the same final diameter (~ 8 nm).

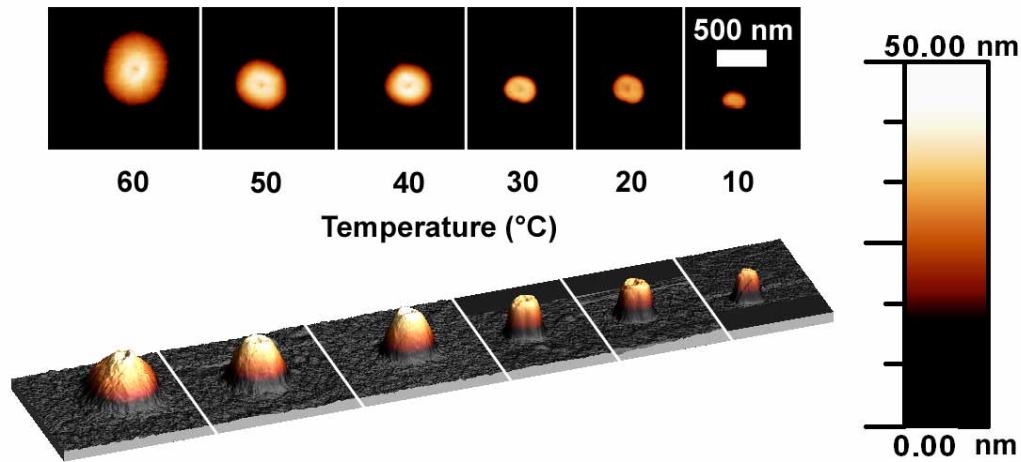


Figure 2. Nanovolcano sizes vary with temperature. Initial diameter is 100 nm for all pores, final diameter 8 nm.

It has previously been observed that pores that have been partially closed at high temperature will open at low temperature and continue closing at high temperature again [2]. Thus, a pore can be kept at approximately constant size by alternating temperatures at which the pore will open and close. Accordingly, to determine the temperature dependence of the instantaneous closing rate, the temperature of a single sample was cycled during closing. The pore was closed at 80 °C to about half its initial area. The temperature was then varied, alternating high (closing) and low (opening) temperatures to maintain a relatively constant area, as shown in Figure 3.

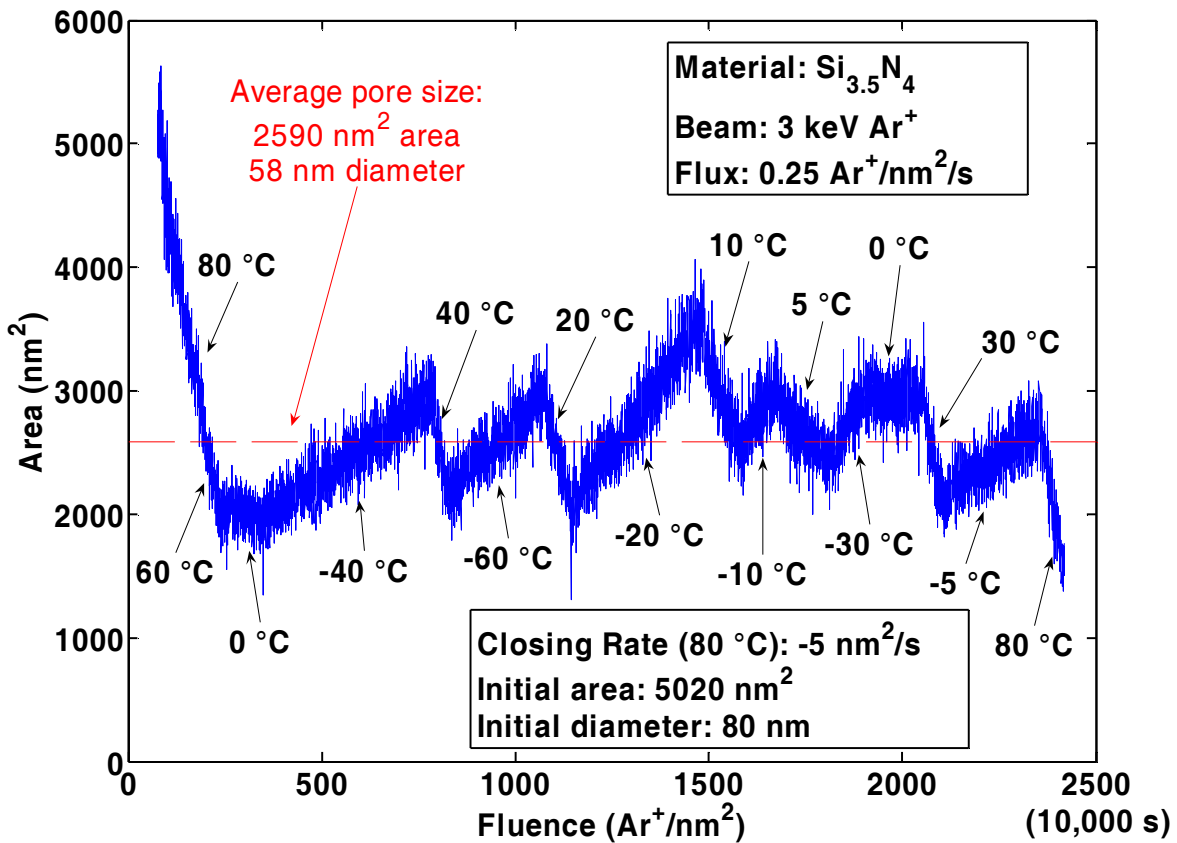


Figure 3. Nanopore ion sculpting signals vs. temperature. The red dashed line indicates the average pore size at which the closing rates in Figure 4 were calculated.

Figure 4 depicts the change in instantaneous slope with temperature, normalized to the closing rate of the pore at 80 °C. The saturation of closing rate at high temperature was first observed very recently by H. B. George and M. J. Aziz in silicon oxide and amorphous silicon nanopores [15], and our data show that the same effect occurs in silicon nitride. The temperature transition seems to indicate the presence of some thermally activated process, but it is not immediately clear why that process should saturate at high temperature. We present an explanation for the saturation behavior later in this paper.

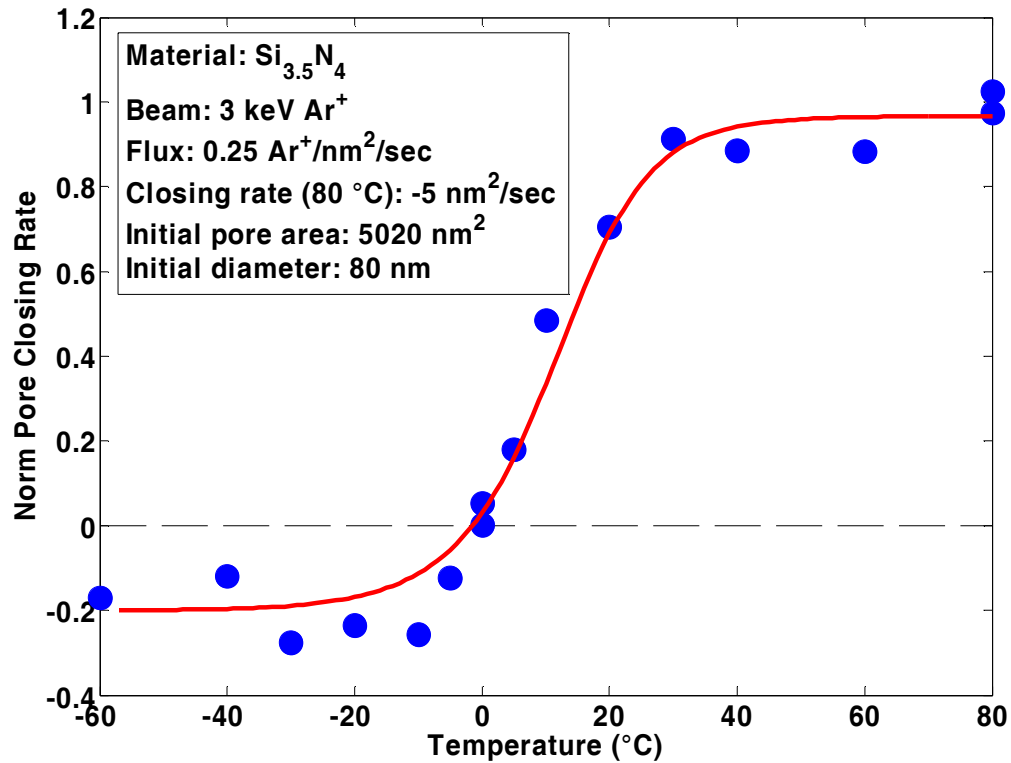


Figure 4. Nanopore ion sculpting rates vary with temperature. Rates are normalized to the area closing rate of the sample at 2590 nm^2 and $80 \text{ }^\circ\text{C}$; negative rates indicate pore opening. The red curve is a fit to the diffusion model (discussed below).

The sensitivity of volcano size to changes in temperature in the transition region allows us to control the shape of the nanopore. Figure 5 shows three nanopores: one was closed at constant temperature, creating a standard nanovolcano; one at two different temperatures, producing a small volcano on top of a large one; and one at gradually decreasing temperatures, yielding a volcano with a smooth transition to the flat surface.

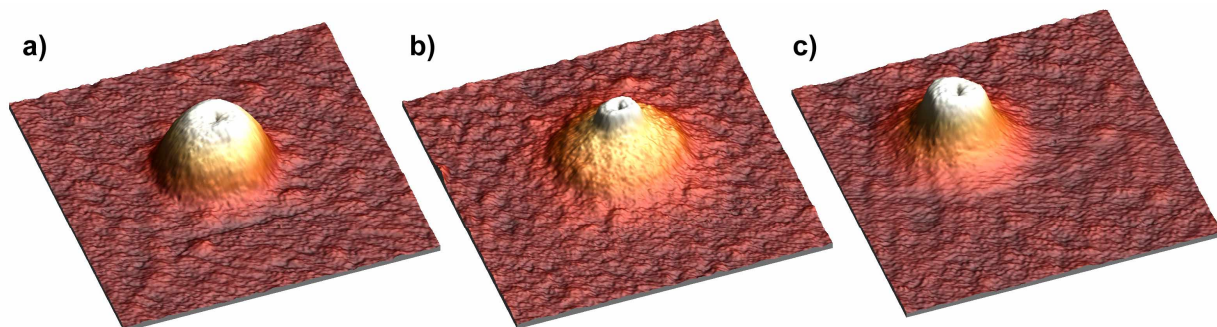


Figure 5. Ion sculpting of nanopores. Nanopores were closed at $0.9 \text{ Ar}^+/\text{nm}^2/\text{sec}$ flux, 100 nm initial diameter, and temperature (a) constant $40 \text{ }^\circ\text{C}$; (b) $60 \text{ }^\circ\text{C}$ to half initial area, then $20 \text{ }^\circ\text{C}$; (c) temperature decreasing from $60 \text{ }^\circ\text{C}$ to $40 \text{ }^\circ\text{C}$ to $20 \text{ }^\circ\text{C}$. Note the "volcano on a volcano" feature in (b). Heights are (a) 32 nm ; (b) 28 nm ; (c) 29 nm above the sputtered surface. All figures have lateral dimensions $1425 \text{ nm} \times 1425 \text{ nm}$.

Initial size dependence

We also investigated the dependence of volcano shape and size on initial pore area. Four nanopores, of nominal initial diameters 100 nm, 200 nm, 400 nm, and 600 nm, were closed completely under identical conditions: temperature 80 °C, flux 0.25 Ar⁺/nm²/sec. The membrane thickness for these pores was 250 nm.

Pores with larger initial diameters form much larger volcanoes (see Figure 6) and thus require more material to shrink the pore by adding another “layer” to the volcano. Assuming mass transport occurs at a constant speed, pores with larger volcanoes should close more slowly at a given area, in this case 4000 nm². Therefore, we expect that for a given instantaneous pore diameter, the closing rate should be slower for pores that started larger. Indeed, we observe that pores with initial diameters larger than 200 nm close more slowly with increasing pore size (see Figure 7 and Figure 8). We do not have an explanation for the drop in closing speed for small initial diameters. In order to solidify the relationship between instantaneous volcano shape and closing rate, future work might include investigating closing at some final diameter and comparing the closing rates at that diameter to the volcano shape at that point.

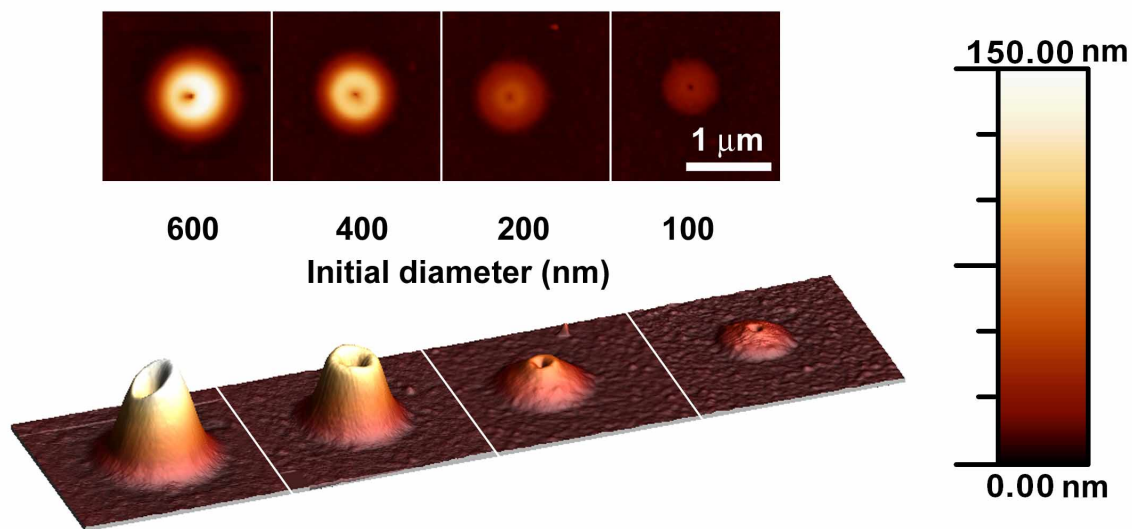


Figure 6. Initial area dependence of volcano size. All pores were closed at 80 °C and a flux of 0.25 Ar⁺/nm²/sec. Initial diameters are nominal.

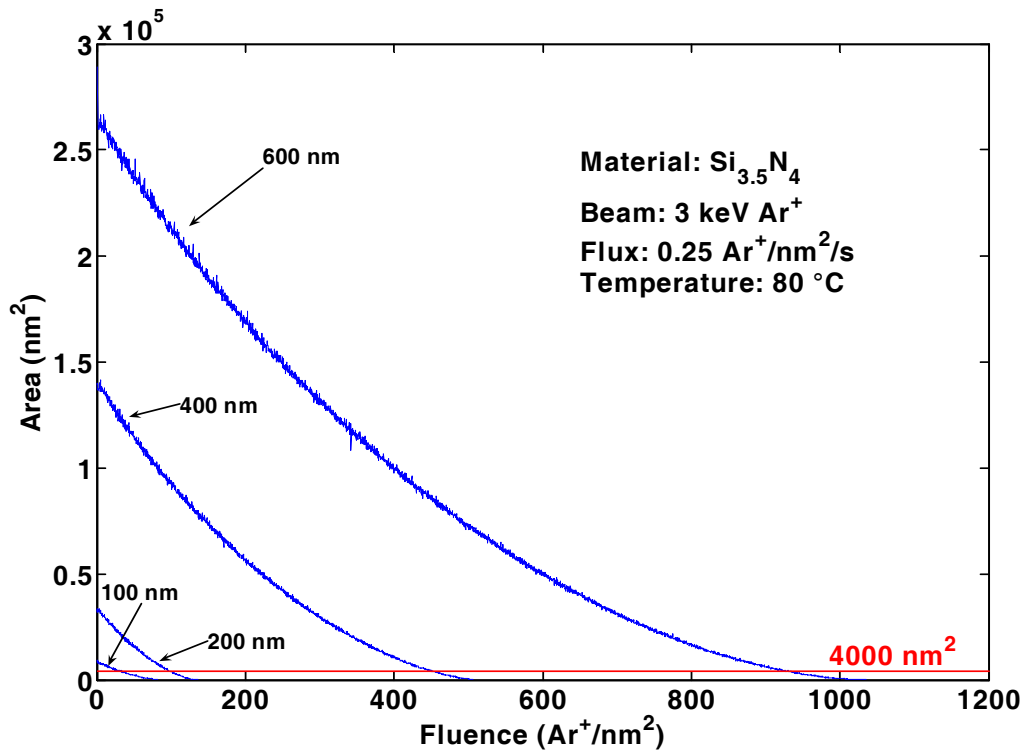


Figure 7. Dependence of ion sculpting behavior on initial nanopore area. Instantaneous slopes are compared at 4000 nm^2 in Figure 8.

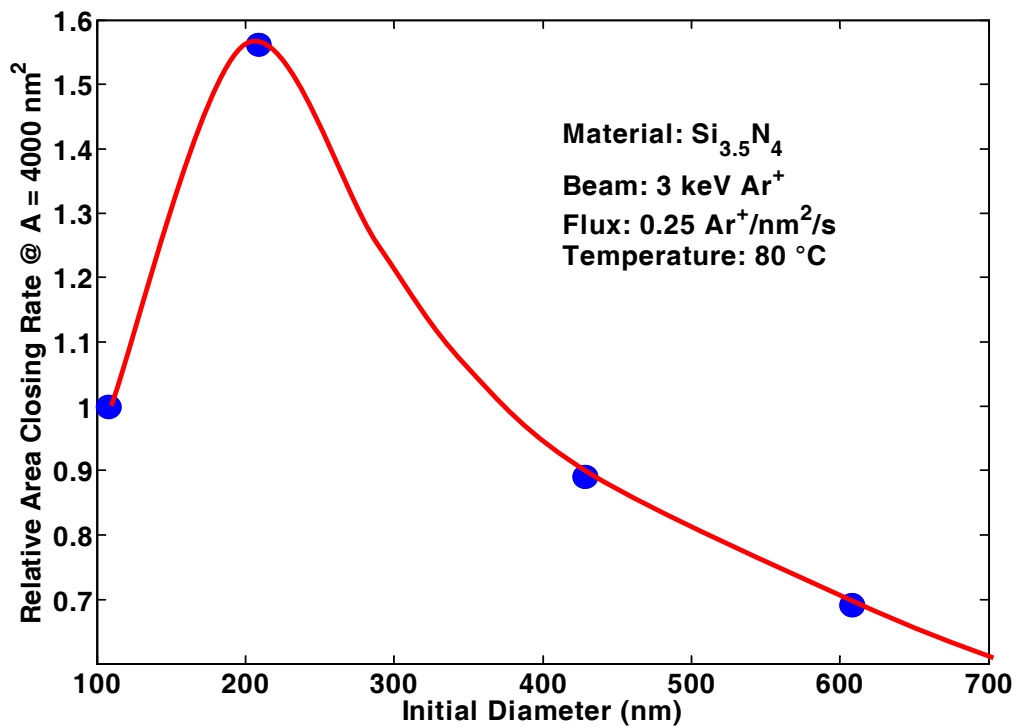


Figure 8. The closing rate at a given area relates to the initial size of the nanopore. The red curve is a guide to the eye. Initial diameters were determined by transmission electron microscopy (TEM).

DISCUSSION

In order to explain pore closing data, various models have been explored. Originally, Li *et. al.* [2] advanced a surface diffusion model in which beam-generated adatoms diffused on the surface until they were captured by surface traps, were immobilized on the pore rim, or were annihilated by the beam. The closing behavior in this model is characterized by the parameter X_m , the average distance an adatom travels before it is trapped or destroyed by the beam, or, equivalently, the size of a depletion region of adatoms around the pore. X_m is given by Equation 1, where l_{trap} is the average distance between surface traps, D is the surface diffusivity, σ is the cross-section for annihilation by the beam, and f is the beam flux.

$$\frac{1}{X_m^2} = \frac{1}{l_{trap}^2} + \frac{\sigma}{D} f \quad (1)$$

The surface diffusion model required modification in order to explain both the accretion of relatively vast amounts of material in the nanovolcanoes and the large distances from which this material appeared to be coming. Mitsui *et. al.* [14], after studying sculpting dynamics of arrays of nanopores, proposed an enhancement to the diffusion process resulting from large electric fields on the surface.

The surface diffusion model also lends itself readily to incorporation of temperature effects. Except for very large pores, the closing rate is proportional to X_m . It is then clear from Equation 1 that an Arrhenius form of the surface diffusivity will yield the sigmoidal behavior observed for the temperature dependence in Figure 4. At high temperatures, where diffusion is significant, the average “lifetime” of an adatom is limited by the trap concentration l_{trap} . At low temperatures, the adatom is diffusion-limited and is more likely to be annihilated by the beam than to fall into a trap or reach the pore periphery. If at reduced temperatures diffusion is thermally suppressed sufficiently, the removal of material from the pore periphery by sputtering dominates, and the pore opens. A typical value for the average trap separation distance is $l_{trap} \sim 65$ nm. The data are not sufficiently precise to extract reliable Arrhenius parameters; we only report that the surface diffusivity varies from about 1 to 10^3 nm²/s over the temperature range of the experiment.

The primary weakness of the surface diffusion model lies in explaining the size and shape of the volcanoes; even the possible enhancement by charging seems inadequate to explain this extraordinary effect in detail. These concerns with the diffusion model have led some to speculate about the contribution of ion-enhanced viscous flow such as that seen for MeV beams [6–8]. In this case the viscosity of the irradiated surface is reduced considerably by the deposition of energy from the beam, causing flow that responds to intrinsic or ion-induced stresses in the surface. It is natural to explain volcanoes as buckling due to stress relief or the result of a collective flow toward the pore. The dependence on initial pore size in this scenario may arise from differences in stress from the initial FIB milling or from the curvature of the pore.

The viscous flow model is difficult to quantify, however, and it is unclear how the observed temperature dependence would arise in such a model, where the important terms are not strongly temperature-dependent at room temperature [10]. The “thermal spike” usually

associated with ion impingement at high energy should not be sensitive to the relatively small changes in surface temperature in our experiments.

CONCLUSIONS

Despite the new data presented in this paper, it is still inconclusive whether surface diffusion, viscous flow, a combination of these, or some other mechanism is responsible for the large matter transport observed in the ion sculpting process. We have introduced constraints on possible models based on temperature and initial size dependence and have demonstrated how these effects can be used to control the shape of a nanovolcano. Understanding the material transport mechanism and its interplay with electric fields and the underlying material properties could greatly aid in the precise fabrication of novel nanostructures.

Ion sculpting of nanopores also presents a powerful platform for studying the effects of keV ion beams on amorphous surfaces. Despite increasing interest in nanofabrication using lab-scale ion beams, a clear microscopic understanding of the interaction of keV ion beams with amorphous surfaces remains elusive. It is difficult to imagine modeling these interactions without elucidating and accounting for the mechanisms that create nanovolcanoes. Any answer to the question of matter transport under ion irradiation, be it individual adatom diffusion, collective flow, or a combination of the two, further influenced by implanted charge, must be able to explain the dramatic presence and morphology of nanovolcanoes.

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