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Iron-Catalyzed 1,4-Hydroboration of 1,3-Dienes

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Iron can adopt formal oxidation states ranging from -II¹ to +VI.² Iron complexes are used as catalysts in synthetic chemistry for carbon–heteroatom³ and carbon–carbon bond forming reactions.⁴ Low-valent iron complexes can catalyze cross-coupling,⁴ cycloisomerization,⁵ and cycloaddition reactions.⁶? We are interested in iron catalysis for the identification of useful, previously inaccessible reaction chemistry and report here C–B bond formation by hydroboration of 1,3-dienes. To our knowledge, there are no other examples of Fe-catalyzed hydroboration reactions of olefinic substrates. The allylborane products are formed regio- and stereoselectively with (*E*)-double bond geometry exclusively and are challenging to access selectively with conventional chemistry.

We have previously reported a 1,4-addition reaction of α -olefins to dienes using an iminopyridine-ferrous chloride⁸ complex with magnesium metal as an in situ reducing agent.⁹ In this Communication, we describe the use of analogous, readily prepared iminopyridine-derived iron complexes as catalysts for the regioselective 1,4-addition of pinacolborane (HBPin) to substituted 1,3-dienes (eq 1).

Allylboranes are versatile intermediates employed in oxidation to allylic alcohols, ¹⁰ allylation to give homoallylic alcohols ¹¹ and amines, ¹² and Suzuki cross-coupling reactions. ¹³ Traditional methods to synthesize allylboranes involve basic main group organometallics, such as Grignard and organolithium reagents, that are incompatible with electrophilic functional groups. ^{11a} While transition-metal-catalyzed hydroboration of olefins has been studied extensively with great success, ¹⁴ hydroboration of dienes to access allylboranes is less established. Palladium(0) catalyzes the 1,4-addition of catecholborane to unfunctionalized 1,3-dienes such as 1,3-pentadiene and isoprene to give (Z)-branched allylboranes, ¹⁵ and Ni^{II}- ¹⁶ and Rh^I-catalyzed ¹⁷ reactions are selective for 1,2-addition.

The synthesis of linear (E)- γ -disubstituted allylboranes is attractive because, for example, they can afford trisubstituted allylic alcohols stereospecifically and add to electrophiles to generate quaternary stereocenters with control of diastereoselectivity. Challenges in the synthesis of linear (E)- γ disubstituted allylboranes via hydroboration of 1,3-dienes include control of chemoselectivity to favor 1,4- over 1,2-addition, control of regioselectivity to favor C-B bond formation at a single diene terminus, and control of stereoselectivity to favor E-olefin geometry. The hydroboration reaction presented herein controls all three types of selectivity. A general method to synthesize linear (E)-γ-disubstituted allylboranes has not been reported previously.18

We observed that hydroboration of myrcene (1) to geranylpinacolborane (2a) was catalyzed by the iminopyridine-iron(II) complex 3·FeCl₂, upon addition of magnesium metal, after 3 h at 23 °C in 80% yield (eq 1). The combination of ligand, ferrous chloride, and magnesium was necessary for catalysis. Pinacolborane generally afforded allylboranes that were stable to

air, water, and chromatography on silica gel; other borolanes, such as those derived from catecholborane, were not stable towards hydrolysis or chromatography on silica gel.¹⁹

Evaluation of different bidentate ligands (see Supporting Information) showed that iminopyridine ligands gave the highest yields for hydroboration. The redox activity of iminopyridine ligands may play a role in effecting efficient catalysis.²⁰ Ligand optimization revealed that variation of the substituent of the imine nitrogen modulates the 1,4-regioselectivity to favor either the branched or the linear isomer (eqs 1 and 2). 1,4-Addition of pinacolborane to myrcene (1) catalyzed by 3·FeCl₂ produced geranylpinacolborane (2a) in 93:7 (2a:2b) regioselectivity and >99:1 E/Z selectivity in 89% overall yield (eq 1). When catalyst 4·FeCl₂ was used, the regioselectivity inverted to afford branched allylborane 2b as the major product in 78% isolated yield (92% yield of combined regioisomers, eq 2). The ligand-controlled regioselectivity is of synthetic value and may be set during migratory insertion (see mechanistic hypothesis, Scheme 1).

Hydroboration of various 1,3-dienes occurred within 4 min to 4 h, depending on the substrate and ligand, and proceeds as efficiently with commodity ferrous chloride (98% purity) as with high-purity ferrous chloride (99.998% purity) as iron source. As shown in Table 1,21 the regioselectivity for 1,4-hydroboration of 2-substituted dienes increased as the size of the 2-substituent increased: isoprene afforded prenylboronate ester 10a with 90:10 regioselectivity; geranylborane 2a was obtained in 93:7 regioselectivity; 2-cyclohexylbutadiene dimethylphenylsilylbutadiene (13) were hydroborated in 94:6 and 99:1 regioselectivity, respectively. The regioselectivity of 1,4addition to 13 could be inverted from 99:1 to < 1:99 by using the iron complex 4·FeCl2, which differs only in the iminopyridine substituent from iron complex 3-FeCl, (entries 6, 7). Both allylboranes 14a and 14b are difficult to synthesize otherwise: silaboration of allenes yields 2-borylallylsilanes.²² In addition to 2-substituted dienes, 2,3-disubstituted diene 5 participated in Fecatalyzed hydroboration to regioselectively give allylborane 6. The 1,4-disubstituted diene 7 was hydroborated efficiently to give allylborane 8. Hydroboration of the 1,2-disubstituted diene (+)nopadiene (15) gave C-B bond formation at the less substituted diene terminus with 98:2 regioselectivity.²³

Table 1. Substrate scope for Fe-catalyzed 1,4-hydroboration of 1,3-dienes.

myrcene

(+)-nopadiene

ÓМе

8

13

R	+ H=B Me	mol% LFeCl ₂ 10 mol% Mg Et ₂ O, 23 °C 4 min - 4 h H R	+ PinB R	Ar = 3	Ar N = 3 Ar B,5-dimethylphenyl	L = 4	iPr iPr
Entry	Diene	Linear Product	Branched Product	L	(Linear: Branched)	E:Z (major)	Yield
1	Me 5	PinB Me 6		4 ^a			91%
2	7	8		4 ^a			83% ^b

Me

PinR

PinB PinB

12a

16a

ÓМе

CO₂tBu

observed upon purification by chromatography on silica gel; yield can be higher if used in situ. See, for example, eq 4.

10b

2b

12b

14b

14b

16b

18b

CH(OMe)₂

3

3

The Fe-catalyzed hydroboration can be performed in the presence of electrophilic functionality, such as the ester in 19, that is incompatible with the basic conditions of traditional Notably, hydroboration allylborane syntheses.11a chemoselective, and 1,4-addition to 1,3-dienes proceeds without hydroboration of isolated olefins such as in 21. We attribute the high chemoselectivity of diene versus olefin hydroboration to the affinity of 1,3-dienes for low-valent iron.²⁴ The Fecatalyzed hydroboration of all dienes investigated is selective for 1,4-addition to produce allylboranes stereo- and regioselectively. 1,2-Addition products could not be detected by ¹H NMR. The major regioisomers 12a-22a were formed with (E)-double bond geometry exclusively, consistent with the proposed mechanism in Scheme 1. Products with trisubstituted double bonds can otherwise be challenging to synthesize stereoselectively, especially when the substituents are similar in size.25

Allylic alcohols with trisubstituted double bonds are substrates for asymmetric catalytic reactions, such as Noyori hydrogenation, ²⁶ isomerization to generate chiral aldehydes, ²⁷ and Sharpless asymmetric epoxidation. ²⁸ All reactions require the alcohol substrate to be stereochemically pure with respect to double bond geometry to attain high levels of enantioselectivity. The presented Fe-catalyzed hydroboration provides ready access

to allylic alcohols with (E)-trisubstituted double bonds in high stereoselectivity (> 99:1, eqs 3 and 4).²⁵ For example, allylic alcohol **23** was synthesized in two steps by hydroboration of (+)-nopadiene (**15**) followed by oxidation to provide only E-isomer **23** in 84% yield over 2 steps.

90:10

93:7

94:6

99:1

< 1:99

98:2

94:6

94:6

> 99:1

> 99:1

> 99:1

> 99:1

> 99:1

> 99:1

92%°

89%

68%^{b,d}

91%b

73%^b

91%^b

71%^b

66%^{b,d}

The Fe-catalyzed reaction can also be used in a one-pot hydroboration-allylation reaction, as shown in eq 5. Benzaldehyde was added, subsequent to 1,4-hydroboration of ester 19, to give a 93:7 mixture of homoallylic alcohols 25a and 25b resulting from the two regioisomers 20a and 20b. Homoallylic alcohol 25a was formed as a single diastereomer, as determined by ¹H NMR spectroscopy. Lactone formation afforded 26 in 85% yield as a single diastereomer. Alcohol 25a and δ-lactone 26 both contain an all-carbon quaternary center; the relative stereochemistry in 25a and 26 is a result of the E-

configuration of the y-disubstituted allylborane generated in the Fe-catalyzed hydroboration.

A preliminary mechanistic analysis led us to propose the catalytic cycle shown in Scheme 1. The deuterium atom from pinacolborane-d was found at the methyl group of the hydroborated product 2a-d exclusively. Selective deuteration is consistent with migratory insertion into either the Fe-B or the Fe-H bond via the iron allyl intermediates 29a and 29b, respectively but cannot distinguish between the two pathways. Proposed compounds 29a and 29b were not observed during catalysis. The turnover-limiting step and the reversibility of the steps of the catalytic cycle are currently unknown and, hence, the ligand-controlled regioselectivity (e.g. entries 6 vs. 7) could be determined during oxidative addition or migratory insertion. When the branched isomer 2b was subjected to the reaction conditions of hydroboration, no linear isomer 2a was observed, which established that at least one step after the regioselectivitydetermining step is irreversible. The selectivity for double bond geometry can be rationalized by the proposed mechanism due to syn migratory insertion to Fe-allyl **29a** or **29b**.

Scheme 1. Proposed mechanism for 1,4-hydroboration.

In conclusion, we report a chemo-, regio-, and stereoselective Fe-catalyzed hydroboration of 1,3-dienes to afford linear (E)- γ disubstituted allylboranes. The iminopyridine-Fe-catalyzed reaction provides access to allylboranes-versatile building blocks-that are challenging to prepare by traditional allylborane syntheses or other known transition metal-catalyzed reactions. The hydroboration reaction demonstrates previously unknown reactivity for iron.

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Supporting Information Available: Detailed experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

- (a) Jonas, K.; Schieferstein, L.; Krüger, C.; Tsay, Y. H. Angew. Chem., Int. Ed. Engl. 1979, 18, 550-551. (b) Ellis, J. E. Organometallics 2003, 22, 3322-3338. (c) Fürstner, A.; Krause, H.; Lehmann, C. W Angew.
- Chem., Int. Ed. **2006**, 45, 440–444. Berry, J. F.; Bill, E.; Bothe, E.; George, S. D.; Mienert, B.; Neese, F.; Wieghardt, K. Science **2006**, 312, 1937–1941.
- (a) Barton, D. H. R.; Gastiger, M. J.; Motherwell, W. B. J. Chem. Soc., Chem. Commun. 1983, 1, 41-43. (b) Chen, M. S.; White, M. C. Science **2007**, 318, 783–787.
- For reviews, see: (a) Leitner, A. Iron-catalyzed Cross-coupling Reactions. In *Iron Catalysis in Organic Chemistry*; Plietker, B., Ed.; Wiley-VCH: Weinheim, 2008; pp 147–176. (b) Sherry, B. D.; Fürstner, A. Acc. Chem. Res. **2008**, 47, 1500–1511. (c) Fürstner, A. Angew. Chem., Int. Ed. 2009, 48, 1364-1367.
- (a) Takacs, J. M.; Anderson, L. G.; Madhavan, G. V. B.; Creswell, M. W.; Seely, F. L.; Devroy, W. F. Organometallics 1986, 5, 2395–2398.
 (b) Fürstner, A.; Martin, R.; Majima, K. J. Am. Chem. Soc. 2005, 127, 12236–12237.
 (c) Fürstner, A.; Majima, K.; Martin, R.; Krause, H.; Kattnig, E.; Goddard, R.; Lehmann, C. W. J. Am. Chem. Soc. 2008, 130, 1992-2004.
- For examples, see: (a) tom Dieck, H.; Dietrich, J. Chemische Berichte-Recueil 1984, 117, 694-701. (b) tom Dieck, H.; Dietrich, J. Angew. Chem., Int. Ed. Engl. 1985, 24, 781-783. (c) Bouwkamp, M. W.; Bowman, A. C.; Lobkovsky, E.; Chirik, P. J. J. Am. Chem. Soc. 2006, 128, 13340–13341.
- For reviews, see: (a) Bolm, C.; Legros, J.; Le Paih, J.; Zani, L. Chem. Rev. 2004, 104, 6217–6254. (b) Enthaler, S.; Junge, K.; Beller, M. Angew. Chem., Int. Ed. 2008, 47, 3317–3321. (c) Correa, A.; Mancheño, O. G.; Bolm, C. Chem. Soc. Rev. 2008, 37, 1108–1117. (d) Iron Catalysis in Organic Chemistry; Plietker, B., Ed.; Wiley-VCH: Weinheim, 2008.
- Similar iminopyridine-FeCl, complexes have been reported: Gibson, V. C.; O'Reilly, R. K.; Wass, D. F.; White, A. J. P.; Williams, D. J. Dalton Trans. 2003, 2824-2830.

- Moreau, B.; Wu, J. Y.; Ritter, T. Org. Lett. 2009, 11, 337–339.
 Brown, H. C.; Zaidlewicz, M. In Organic Syntheses Via Boranes, Vol. 2; Aldrich Chemical Company: Milwaukee, 2001; pp 118–120.
 (11) (a) Yamamoto, Y.; Asao, N. Chem. Rev. 1993, 93, 2207–2293. (b) Bubnov, Y. N. Pure Appl. Chem. 1987, 59, 895–906. (c) Brown, H. C.; Taidley, M. M. Markey, V. M. P. Vict. 2011. Zaidlewicz, M. In Organic Syntheses Via Boranes, Vol. 2; Aldrich Chemical Company: Milwaukee, 2001; pp 189–194 and pp 207–216. (d) Kennedy, J. W. J.; Hall, D. G. *Angew. Chem., Int. Ed.* **2003**, 42, 4732-4739.
- (12) For recent examples, see: (a) Sugiura, M.; Hirano, K.; Kobayashi, S. J. Am. Chem. Soc. 2004, 126, 7182–7183. (b) Solin, N.; Wallner, O. A.; Szabó, K. J. Org. Lett. 2005, 7, 689–691. (c) Lou, S.; Moquist, P. N.; Schaus, S. E. J. Am. Chem. Soc. 2007, 129, 15398–15404.
 (13) Sebelius, S.; Olsson, V. J.; Wallner, O. A.; Szabó, K. J. J. Am. Chem. Soc. 2006, 128, 8150–8151.
- (14) For reviews of transition-metal-catalyzed hydroboration, see: (a) Burgess, K.; Ohlmeyer, M. J. *Chem. Rev.* **1991**, *91*, 1179–1191. (b) Beletskaya, I.; Pelter, A. *Tetrahedron* **1997**, *53*, 4957–5026.
- Satoh, M.; Nomoto, Y.; Miyaura, N.; Suzuki, A. Tetrahedron Lett. **1989**, 30, 3789–3792
- Zaidlewicz, M.; Meller, J. Tetrahedron Lett. 1997, 38, 7279–7282. Matsumoto, Y.; Hayashi, T. Tetrahedron Lett. 1991, 32, 3387–3390.
- (18) For a Pd-catalyzed synthesis of geranylpinacolborane from geraniol, see: (a) Dutheuil, G.; Selander, N.; Szabó, K. J.; Aggarwal, V. K. Synthesis 2008, 14, 2293–2297. Pt(0)-catalyzed diboration of isoprene gave a 2:1 isoprene:B₂(Pin)₂ adduct: (b) Ishiyama, T.; Yamamoto, M.; Miyaura, N. *Chem. Commun.* **1996**, 2073–2074. For an example of 1,4diboration of 1,3-dienes to access (Z)-allylic diboranes, see: (c) Ballard, C. E.; Morken, J. P. Synthesis 2004, 1321-1324.
- (19) Alkylboranes such as 9-BBN and diisopinocampheylborane were ineffective hydroborating agents under the reaction conditions shown in egs 1 and 2
- (20) Lu, C.; Bill, E.; Weyhermüller, T.; Bothe, E.; Wieghardt, K. J. Am. Chem. Soc. **2008**, 130, 3181–3197.
- 2,3-Dimethyl-1,3-butadiene was used as an additive to stabilize lowvalent iron generated from reduction of the ferrous chloride complex.
- Suginome, M.; Ohmori, Y.; Ito, Y. J. Organomet. Chem. 2000, 611, 403-413.
- (23) For Fe-catalyzed hydroboration of the 1-substituted diene 1,3decadiene, see Supporting Information. (24) Knölker, H. J. Chem. Soc. Rev. 1999, 28, 151–157.
- (a) Yu, J. S.; Kleckley, T. S.; Wiemer, D. F. *Org. Lett.* **2005**, *7*, 4803–4806. (b) Bissada, S.; Lau, C. K.; Bernstein, M. A.; Dufresne, C. *Can. J. Chem.* **1994**, *72*, 1866–1869. (c) Bellucci, C.; Gualtieri, F.; Scapecchi, S.; Teodori, E.; Budriesi, R.; Chiarini, A. *Il Farmaco* 1989, 44, 1167–1191. (d) For a stereoselective synthesis of trisubstituted allylic alcohols, see: Langille, N. F.; Jamison, T. F. *Org. Lett.* **2006**, 8,
- Takaya, H.; Ohta, T.; Sayo, N.; Kumobayashi, H.; Akutagawa, S.; Inoue, S.; Kasahara, I.; Noyori, R. J. Am. Chem. Soc. 1987, 109, 1596-
- Tanaka, K.; Qiao, S.; Tobisu, M.; Lo, M. M.-C.; Fu, G. C. *J. Am. Chem. Soc.* **2000**. *122*, 9870–9871.
- (28) Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. 1980, 102, 5974-5976.

Iron Catalysis

Chemo-, regio-, and stereoselective 1,4-hydroboration

(E)-isomer only

A chemo-, regio-, and stereoselective iron-catalyzed 1,4-hydroboration of dienes to synthesize γ -disubstituted allylboranes was developed. 1,4-Hydroboration of 2-substituted dienes forms allylborane products with (*E*)-trisubstituted double bonds exclusively.